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CHEMISTRY: A Study of Matter © 2004, GPB 6.28 Classifying Reactions 1. Synthesis: • Two or more substances combine to form a more complex substance. • $A + B = AB$ • $Fe(s) + S(s) = FeS(s)$ • $H_2O(l) + SO_3(g) = H_2SO_4(aq)$ 2.

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Note Taking Guide: Episode 1002 Name _____

Note Taking Guide: Episode 703 Name _____ CHEMISTRY: A Study of Matter © 2004, GPB 7.24 Molecular Formula • represents the _____ number of _____ of each

7-24,25 Note Taking Guide EP 703 - Georgia Public Broadcasting

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Chemistry Matters Physics in Motion Virtual Field Trips ... Unit 4A Note Taking Guide & Segment Questions (172.64 KB) Cleaning Pennies with Taco Sauce (1.21 MB) Chemistry Matters Segments. ... Georgia Public Broadcasting. 260 14th St. NW Atlanta, GA 30318 United States (404) 685-2400 In Atlanta

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Chemistry & Physics | Georgia Public Broadcasting

Note Taking Guide: Episode 1501 Name _____ CHEMISTRY: A Study of Matter © 2004, GPB 15.1 Nucleons: • are particles occupying the _____ • consist of ___ charged _____ and _____ • have almost 2000times the _____ of _____ • are made up of _____ and _____ Nuclear Notation: 27

15-01,02 Note Taking Guide Ep 1501

Title: Microsoft Word - 1-01,02-Note Taking Guide Ep 101.doc Author: Brent White Created Date: 7/12/2005 1:26:34 PM

Note Taking Guide: Episode 101 Name Scientific Method

Note Taking Guide: Episode 701 Name _____ CHEMISTRY: A Study of Matter © 2004, GPB 7.1 Lab results: 1 doz grains of rice = _____g (Use this fact as a conversion

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Chemistry 403: Trends in the Periodic Table - GPB Video

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Note Taking Guide Episode 701 Note Taking Guide: Episode 701 Name: Kelly R. Lab results: 1 doz grains of rice = 0.3 g (Use this fact as a conversion factor.)? grains of rice = 1.94 g Avogadro's Number - the mole = the number 6.022×10^{23} Molar Mass the mass of one mole of any pure substance

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