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This new volume in the Postgraduate Chemistry Series provides a thorough overview of the principles and uses of synthetic organic photochemistry. Appropriate at postgraduate and research level it will also serve as a reference for more experienced workers.

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Organic photochemistry encompasses organic reactions that are induced by the action of light. The absorption of ultraviolet light by

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organic molecules often leads to reactions. In the earliest days, sunlight was employed, while in more modern times ultraviolet lamps are employed. Organic photochemistry has proven to be a very useful synthetic tool. Complex organic products can be obtained simply.

~~Organic photochemistry — Wikipedia~~

Indeed, one of the classic photochemical reactions of organic chemistry is the formation of

1,1,2,2-tetraphenyl-1,2-ethanediol (3, benzopinacol) by the action of light on a

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Practice of diphenylmethanone (2, benzophenone) in isopropyl alcohol. The yield is quantitative.

~~28.3: Organic Photochemistry — Chemistry LibreTexts~~

Abstract Examines the electronics of the simple carbonyl group, the effect of structure on photoreduction of benzophenones and acetophenones, non-conjugated unsaturated ketones, the photochemistry of cyclic enones, and cross-conjugated cyclohexadienones.

~~Photochemistry of organic compounds. II,~~

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Typical absorption range of some important classes of organic compounds: Simple alkene 190 - 200 nm . Acyclic diene 220 - 250 nm . Cyclic diene 250 - 270 nm . Styrene 270 - 300 nm . Saturated ketones 270 - 280 nm . α,β -Unsaturated ketones 310 - 330 nm . Aromatic ketones/aldehydes 280 - 300 nm

~~Photochemistry — Organic Syntheses with Light~~

• The first law of photochemistry, the Grotthuss-Draper law, states that light must be absorbed by a compound in order for a photochemical reaction to take place. • The

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second law of photochemistry, the Stark-Einstein law, states that for each photon of light absorbed by a chemical system, only one molecule is activated for subsequent reaction. This "photoequivalence law" was derived by ...

~~Photochemistry — Michigan State University~~
Photochemistry is the branch of chemistry concerned with the chemical effects of light. Generally, this term is used to describe a chemical reaction caused by absorption of ultraviolet, visible light or infrared radiation. In nature, photochemistry is of

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immense importance as it is the basis of photosynthesis, vision, and the formation of vitamin D with sunlight. Photochemical reactions proceed differently than temperature-driven reactions. Photochemical paths access high energy intermediates th

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Organic mechanoluminescence (ML) compounds have experienced breakthrough developments in recent years, with ML being discovered in many kinds of organic compounds. Accordingly, the ML composition is becoming more

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